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# Synthesis of Potentially Bioactive Hydrazones of Pyrimidine and 1,3,5-Triazine

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## ARTICLE INFO

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### ABSTRACT

The reaction of the pyrimidine and 1,3,5-triazine hydrazides with various aldehydes formed corresponding hydrazones. Synthesized compounds are endowed with potential bioactivity. In laboratory-vegetation screening, the obtained compounds showed a pronounced stimulating effect on plant growth. The activity of the tested compounds varied in the range of 46–88 % as compared to heteroauxin. Compounds that showed an activity above 70 % in the experiment will be selected for deeper study and further field tests.

#### Introduction

Most pharmaceutical preparations that mimic biologically active natural substances consist of heterocyclic scaffolds. Hydrazides and their derivatives, in particular hydrazones, are often used as starting compounds to obtain new heterocyclic structures. Hydrazine derivatives occupy a special place in the chemotherapy of tuberculosis (Mashkovskiy, 2019). Isonicotinic acid hydrazide (isoniazid) has been used in medical practice for more than half century and has not lost its significance to this date. Based on this, ftivazide, saluside, metazide and other modified analogs with improved pharmacological properties have been obtained. In medical practice, antidepressants iproniazid and nialamide (monoamine oxidase inhibitors) are widely used (Mashkovskiy, 2019). Studies of new hydrazine derivatives are ongoing, in which compounds with antimicrobial (Masunari and Tavaris, 2007, Loncle, et al., 2004, Küçükgüzel, et al., 2003, Vicini, et al., 2002), anti-inflammatory (Todeschini, et al., 1998), anti-tuberculosis (Bijev, 2006), antitumor (Gürsoy and Ulusoy-Güzeldemirci, 2007), antimalarial (Gemma, et al., 2006), antidepressant (Ergenç, et al., 1998) and anticonvulsant (Ragavendran, et al., 2007) activities have been revealed. Some hydrazide derivatives, particularly, herbicides (benquinox, saijunmao, phenoxyaryl hydrazides of nicotinic acid) (http://www.alanwood.net/ pesticides/) and also the compounds with plant growth stimulating activity (Gomktsyan, et al., 2012, Yengoyan, et al, 2017, Pivazyan, et al., 2019, Shainova, 2019) are used in agriculture.

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### Materials and methods

We have developed an efficient method for obtaining hydrazine ( $\alpha$ -methylhydrazino)-sym-triazines (Dovlatyan, et al., 1989).

Studies have shown that these compounds have herbicidal properties (Dovlatyan, et al., 1981, Dovlatyan, et al., 1980,1986) and some of their salts have fungicidal effect (Dovlatyan, 1986).

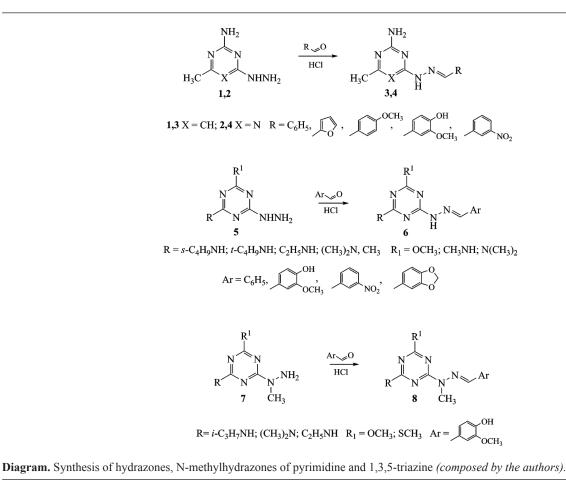
In order to expand the range of biologically active compounds, a number of compounds were interacted with aldehydes in water and appropriate hydrazones were obtained in room temperature.

#### **Results and discussions**

The targeted hydrazones, N-methylhydrazones of pyrimidine and 1,3,5-triazine were synthesized upon the reaction of the corresponding hydrazides and various aryl aldehydes in hydrochloric acid solution (Diagam). The obtained hydrazones can exist as two possible E and

Z structural isomers. However, according to the data of <sup>1</sup>H and <sup>13</sup>C NMR spectra, the sterically more beneficial E-isomer is predominant, and only signals of this isomer are observed in the NMR spectra. The obtained compounds were subjected to laboratory vegetation tests to datarmine harbigidal fungicidal growth regulating

to determine herbicidal, fungicidal, growth-regulating properties. Almost all of the compounds obtained showed a stimulating effect on plant growth. The experiments were carried out on seeds and seedlings of common beans (Phaseolus vulgaris L.). The effect of aqueous suspensions of compounds at the concentrations of 25 and 50 mg/L on seed viability, germination, and seedling growth was studied. These data were compared with the effect of heteroauxin solutions of the same concentrations. The activity of the tested compounds varied in the range of 46-88 % as compared to heteroauxin. In some cases, the growth-stimulating effect of solutions with a lower concentration turned out to be stronger than that of more concentrated solutions. Compounds that showed an activity above 70 % in the experiment will be selected for deeper study and further field tests using their solutions with concentrations less than 25 mg/L.



#### Experimental

<sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded at 30°C on a Varian Mercury-300 NMR spectrometer (300 and 75 MHz, respectively) in a mixture of solvents CCl4–DMSO-*d6* (3:1); TMS was used as an internal standard. The reaction progress and the purity of the obtained compounds were monitored by TLC on Silufol UV-254 plates; benzene–hexane, 2:1, mixture was used as an eluent. Elemental analysis was performed on a Eurovector EA3000 CHNS analyzer. Melting points were determined by the capillary method and were not corrected.

#### General procedure

To the mixture of compounds 1,2,5 or 7 (10 mmol), 10 mL of water and 1,5 mL of HCl (36 %), 10 mmol of various aldehydes were added and the reaction mixture was stirred at 20 °C for 24 h. After adding water (10-15 mL) the precipitate was filtered off and dried.

#### **Compounds examples**

(*E*)-4-(2-benzylidenehydrazinyl)-2-amino-6-methylpyrimidine: Compound yield - 82 %, m.p. 255°C (decomp). <sup>1</sup>H NMR,  $\delta$ : 2.40 (s, 3H, CH<sub>3</sub>); 6.61 (s, 1H, CH-pyrim.); 7.34-7.72 (m, 5H, C<sub>6</sub>H<sub>5</sub>); 7.80 (br.s, 2H, NH<sub>2</sub>); 8.31 (s, 1H, CH=N); 13.54 (br.s, 1H, NH). <sup>13</sup>C NMR,  $\delta$ : 18.4, 95.5, 126.9, 128.1, 129.6, 133.6, 147.1, 153.6, 155.5, 162.7. Anal.calc.: Found: C, 63.29; H, 5.67; N, 30.93. C<sub>12</sub>H<sub>13</sub>N<sub>5</sub>. Calcul.: C, 63.42; H, 5.77; N, 30.82.

#### (*E*)-4-(2-(4-methoxybenzylidene)hydrazinyl)-2-amino-6methylpyrimidine:

Compound yield - 88 %, m.p. 170-172°C. <sup>1</sup>H NMR,  $\delta$ : 2.26 (s, 3H, CH<sub>3</sub>); 3.82 (s, 3H, OCH<sub>3</sub>); 6.36 (br.s, 2H, NH<sub>2</sub>); 6.41 (s, 1H, CH-pyrim.); 6.85-7.61 (m, 4H, C<sub>6</sub>H<sub>4</sub>); 8.02 (s, 1H, CH=N); 11.04 (br.s, 1H, NH). <sup>13</sup>C NMR,  $\delta$ : 21.6, 54.6, 92.3, 113.5, 127.1, 127.7, 142.8, 159.8, 160.1, 161.0, 162.3. Anal.calc.: Found: C, 60.74; H, 5.95; N, 27.40. C<sub>13</sub>H<sub>15</sub>N<sub>5</sub>O. Calcul.: C, 60.69; H, 5.88; N, 27.22.

#### (*E*)-4-(2-(furan-2-ylmethylene)hydrazinyl)-2-amino-6methylpyrimidin:

Compound yield - 83%, m.p. 260 °C (decomp). <sup>1</sup>H NMR,  $\delta$ : 2.39 (s, 3H, CH<sub>3</sub>); 6.52-6.54 (d×d, J<sub>1</sub>=3.4 Hz, J<sub>2</sub>=1.8 Hz, 2H, CH=CH fur.); 6.81 (d, J<sub>1</sub>=3.4 Hz, 1H, CH-fur.); 7.64 (s, 1H, CH-pyrim.); 7.76 (br.s, 2H, NH<sub>2</sub>); 8.19 (s, 1H, CH=N); 12.20 (br.s, 1H, NH). <sup>13</sup>C NMR,  $\delta$ : 18.4, 93.0, 111.6, 113.0, 136.8, 144.3, 149.0, 153.6, 155.4, 162.5. Anal.calc.: Found: C, 55.37; H, 5.02; N, 32.43. C<sub>10</sub>H<sub>11</sub>N<sub>5</sub>O. Calcul.: C, 55.29; H, 5.10; N, 32.24;

# *(E)-4-(2-benzylidenehydrazinyl)-2-amino-6-methyl-1,3,5-triazine:*

Compound yield - 81 %, m.p. 280-282 °C. <sup>1</sup>H NMR,  $\delta$ : 2.28 (s, 3H, CH<sub>3</sub>); 6.80 (br.s, 2H, NH<sub>2</sub>); 7.25-7.73 (m, 5H, C<sub>6</sub>H<sub>5</sub>); 8.10 (s, 1H, CH=N); 11.08 (br.s, 1H, NH). <sup>13</sup>C NMR,  $\delta$ : 24.0, 126.4, 127.9, 128.4, 134.7, 143.4, 173.5. Anal.calc.: Found: C, 57.96; H, 5.25; N, 36.96. C<sub>11</sub>H<sub>12</sub>N<sub>6</sub>. Calcul.: C, 57.88; H, 5.30; N, 36.82.

#### (*E*)-4-(2-(4-methoxybenzylidene)hydrazinyl)-2-amino-6methyl-1,3,5-triazine:

Compound yield - 83%, m.p. 248-250 °C. <sup>1</sup>H NMR,  $\delta$ : 2.20 (s, 3H, CH<sub>3</sub>); 3.81 (s, 3H, OCH<sub>3</sub>); 6.50 and 7.17 (br.s, 2H, NH<sub>2</sub>); 6.84-7.63 (m, 4H, C<sub>6</sub>H<sub>4</sub>); 8.00 (s, 1H, CH=N); 10.74 (br.s, 1H, NH). <sup>13</sup>C NMR,  $\delta$ : 24.4, 54.7, 113.7, 127.4, 127.7, 128.1, 142.7, 159.8, 163.7; 166.5. Anal.calc.: Found: C, 55.72; H, 5.38; N, 32.38. C<sub>12</sub>H<sub>14</sub>N<sub>6</sub>O. Calcul.: C, 55.80; H, 5.46; N, 32.54.

# (*E*)-4-((2-(4-(ethylamino)-6-methoxy-1,3,5-triazin-2-yl)-2-methylhydrazono)methyl)-2-methoxyphenol:

Compound yield - 67%, m.p. 203-205 °C. <sup>1</sup>H NMR,  $\delta$ : 1.29 (t, J=7.2, 3H, NCH<sub>2</sub><u>CH<sub>3</sub></u>); 3.50 (q, J=7.2, 2H, N<u>CH<sub>2</sub></u>CH<sub>3</sub>); 3.64 (s, 3H, NCH<sub>3</sub>); 3.98 (s, 3H, OCH<sub>3</sub>); 4.00 (s, 3H, OCH<sub>3</sub>); 6.75-7.97 (m, 3H, C<sub>6</sub>H<sub>3</sub>); 8.14 (s, 1H, CH=N); 10.62 (br.s, 1H, OH). <sup>13</sup>C NMR,  $\delta$ : 13.7, 30.2, 35.5, 55.2, 56.5, 110.7, 114.5, 124.1, 124.8, 148.0, 148.1, 155.1, 157.2, 168.4. Anal.calc.: Found: C, 54.14; H, 6.01; N, 25.41. C<sub>15</sub>H<sub>20</sub>N<sub>6</sub>O<sub>3</sub>. Calcul.: C, 54.21; H, 6.07; N, 25.29.

#### Conclusion

Based on the aforestated methods and examples, it can be summed up that a series of pyrimidine and 1,3,5-triazine hydrazides derivatives were synthesized, which, upon biological screening, showed a pronounced stimulating effect on plant growth. The most active ones have been selected for deeper research and subsequent field trials.

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